

Download Ideal Gas Laws Ws Answers

Ideal Gas Law Worksheet $PV = nRT$. Use the ideal gas law, “PerV-nRT”, and the universal gas constant $R = 0.0821 \text{ L}\cdot\text{atm} / (\text{K}\cdot\text{mole})$ to solve the following problems: $\text{K}\cdot\text{mol}$. If pressure is needed in kPa then convert by multiplying by $101.3 \text{ kPa} / 1 \text{ atm}$ to get $R = 8.31 \text{ kPa}\cdot\text{L} / (\text{K}\cdot\text{mole})$

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